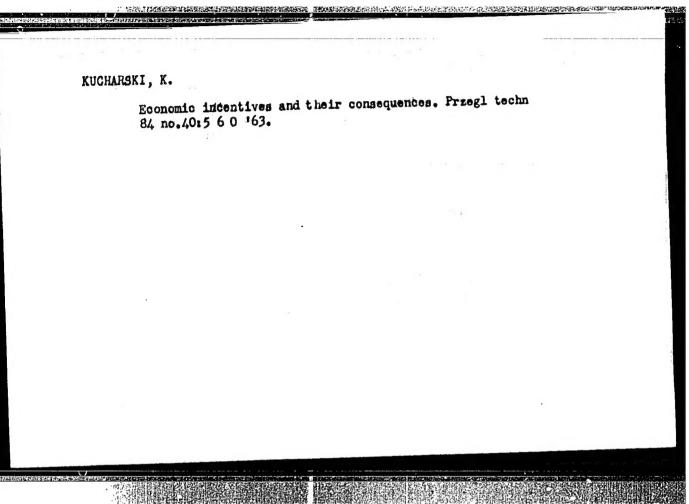
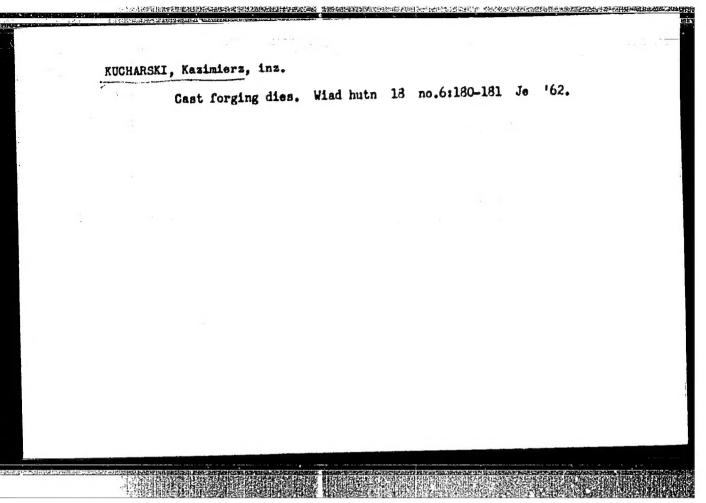
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KUCHARSKI, Josef Kasimiers; SZYSZKO, Bosena; MAHDAROWICZ, Cseslaw

A case of laboratory tetanus. Polski tygod.lek.15 no.7:263-265 15 F '60.





KUCHARSKI, Kazimiers, inz.

Durability of thin piston rods of steam-air harmers.

Wiad hut 18 no.7/8:221-223 Jl-Ag '62.

KUCHARSKI, Kazimierz, inz.

Export production of locomotive turnouts. Przegl techn no.6: 5 10 F '63.

1. Huta im. K. Swierczewskiego, Zawadskie.

KUCHARSKI, Kazimiera, ins.

How to increase the number of regenerations in forging dies. Wiad but 19 no.1:24-25 Ja '63.

Iron casting industry in Opole Prov no.25:8. Je 162.	ince. Przegl techn
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WUSATOWSKI, Zygmunt; KRAWCZYK, Ryszard; KUCHARSKI, Kazimierz

High draught hot rolling of M St 7 steel. Metal i odlew no.7:161-205 '61.

1. Politechnika Slaska, Glivice.

KUCHARSKI, Kazimierz, mgr inz.

The die card. Wiad hut 19 no.12:359-361 D\*63.

## "APPROVED FOR RELEASE: 03/13/2001 CI

CIA-RDP86-00513R000827030002-9

L 45851-66 EWP(j)/T IJP(c) WW/RM

ACC NR: AP6029091 (4) SOURCE CODE: GE/0004/66/000/006/0330/0332

AUTHOR: Rabek, T. I.; Kucharski, M.; Skowronski, T.; Wojaczynska, Maria; Zuchowska, Danuta

ORG: Institute of Chemistry of Plastics and Technology, Polytechnic Institute, Wroclaw (Institut fur Chemie und Technologie der Plaste der Technischen Hochschule)

TITLE: Copolymerization of butadiene with styrene to products of low molecular weight

SOURCE: Plaste und Kautschuk, v. 13, no. 6, 1966, 330-332

TOPIC TAGS: mixed polymerization, butadiene, oligomer, polymer cross linking

ABSTRACT: The following optimum conditions are suggested for the polymerization of butadiene with styrene (with the composition given in parts by weight): 75 butadiene, 25 styrene, 5 sodium, 100 benzene, 30 dioxane, 0.2 sodium isopropylate. Polymerization temperature ranges from 60 to 70C; the reaction time is 8 hr.; molecular weight of the copolymer is 307 and the temperature jump at cross-linking is 20C. The copolymer yield is a function of temperature; it is lower

Card 1/2

L 45851-66 ACC NR: AP6029091

between 50 and 70C in an inert solvent than at temperatures above 70C. The addition of ether and dioxane results in higher styrene quantity in the copolymer. The effect is higher with dioxane (18 to 27%) than with ether (14 to 16%). The increase of ether and dioxane from 10 to 50 (pt. wt.) lowers the molecular weight. The cross-linking capacity of the copolymer is independent of the amount of styrene in it (within the range from 20 to 28%) and of the molecular weight, although it depends on the share of 1.2-structure in the copolymer. The content of the 1.2-structure is markedly affected by dioxane during copolymerization. The optimum cross-linking parameters are: a 120C temperature, and a 1-hr reaction time followed by heating at 150C for 5 hr. The product obtained represents a cross-linked polymer, insoluble in organic solvents and nonexpandable. Orig. art. has: 10 figures and 3 tables. [Based on author's abstract]

SUB CODE: 07, 11/ SUBM DATE: .08Jun65/ SOV REF: 003/ OTH REF: 010/

Card 2/2 JS

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Coprecipitation of phosphates and sulfates with hydroxides. Rocz chemii 34 no.5:1265-1274 60. (EEAI 10:9)

1. Radiochemical Laboratory, Institute of Nuclear Research, Warszawa.

(Precipitation(Chemistry)) (Phosphates) (Sulfates) (Hydroxides)

ACCESSION NR: AP4038571

0/0025/64/000/001/0038/0040

AUTHOR: Plejewski, R.; Kucharski, M.

TITLE: New method for producing carrier-free H sub 2 8 sup 35 0 sub 4 on a production scale

SOURCE: Kernenergie, no. 1, 1964, 38-40

TOPIC TAUS: radioisotope, production, H sub 2 S sup 35 0 sub h, carrier-free, separation, neutron, radiation, reactor, KCE

ARSTRACT: S<sup>35</sup> is obtained by neutron irradiation of KCl in a reactor. Two chromatographic columns are used to separate S<sup>35</sup> from the irradiated target. The first is charged with Al<sub>2</sub>O<sub>3</sub> and serves to separate sulfur from P<sup>32</sup>, Cl<sup>36</sup>, Cl<sup>36</sup>, K<sup>42</sup>, and the main portion of KCl. The second one is charged with a cation exchanger in H-1 form to separate sulfur from all cationic impurities including K<sup>4</sup>-rosidues, and assures the presence of sulfur as H<sub>2</sub>SQ<sub>4</sub>. The column parameters are selected so that a single application of the two columns suffices. Carrier-free, S<sup>35</sup>-tagged H<sub>2</sub>SO<sub>4</sub> in 10<sup>-2</sup> = 10<sup>-3</sup> n HCl is obtained of purity greater than 99%. The specific activity is 35 mc/ml, and the pH is 2-3. The radioisotope may be used for medical

Card 1/2

ACCESSION NR: AP4038571  purposes. "The authors thank S. Lisicki for his valuable support furing the experimental work." Original article has: 3 figures and 2 tables.								
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S/081/62/000/001/038/067 B168/B101

AUTHORS: Kucharski, M., Plejewski, R.

TITLE: A method of producing carrier-free sulfur 535 in the form of

H253504

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 1, 1962, 329, abstract 1K5 ([Referat.] Inst. badań jądrow. PAN, no. 192/XIII, 1960)

TEXT: Radioactive  $3^{35}$  is produced on a commercial scale in the form of  $H_2SO_4$ . In this case the  $H_2SO_4$  is obtained without carrier by irradiation of a KCl target with a neutron flux  $(10^{13} \text{ neutr/cm}^2 \cdot \text{sec})$  for a period of four weeks. The main product is obtained from the reaction  $Cl^{35}(n,p)S^{35}$ . In addition to  $S^{35}$  irradiation produces  $K^{42}$ ,  $P^{32}$ ,  $Cl^{36}$ , and  $Cl^{38}$ ; of these the sulfur and phosphorus immediately oxidize and take the form of ions  $PO_4^{3-}$  and  $SO_4^{2-}$ , whilst the  $K^{42}$  and  $Cl^{38}$ , which have a very short half-life, virtually disappear. After the irradiated target has been Card 1/2

A method of producing...

S/081/62/000/001/038/067 B168/B101

dissolved, the PO<sub>4</sub> is precipitated with Fo(OH)<sub>3</sub> whilst the potassium ions are removed by passing the solution through an ion-exchange column. The chlorine is removed by distillation under reduced pressure, and the organic impurities are oxidized by the addition of 3, H<sub>2</sub>O<sub>2</sub>. Thanks to its chemical and radiochemical purity the S<sup>35</sup> can be used for medical purposes and for the synthesis of labelled products. The technology of H<sub>2</sub>S<sup>35</sup>O<sub>4</sub> production is described. 12 references. [Abstracter's note:

Card 2/2

 MYSAKOWSKA, Helena: Zaluska, Stanislawa; GRODZKI, Stanislaw; KUCHARSKI, Ryszard, PIETRON, Eugeniusz

Clinical forms of pulmonary tuberculosis in women and men from rural and urban environments. Gruslica 27 no.11:1153-1163 N \*59.

1. Z Kliniki Gruslicy Pluc A.M. w Lublinie. Kierownik: doc.dr.
H. Hysakowska.

(TUBERCULOSIS PULMONARY epidemiol.)

KWINKUWA, Agnieszka; LYPACZEWSKA, Joanna; KUCHARSKI, Ryssard; KUCHTA, Jan; KWIT, Wladyslaw; ROPEK, Mieczyslaw

Considerations on the work of anti-tuberculosis dispensaries according to the analysis of records of patients under observation no less than 4 years. Gruslica 27 no.11:1165-1172 H '59.

1. Z Poradni Przeciwgrusliczych: Instytutu Gruslicy w Warszawie, Wojewodzkiej Centralnej w Kublinie, Miejskiej w Walbrzychu i Powiatowej w Chrzanowie.

(TURERCULOSIS hosp.& clinics)

KWIT, Wladyslaw; STASIAK, Eugeniusz; MICH, Jan; KUCHARSKI, Ryszard

Control of human and cattle tuberculosis in the Lublin Region.

Cruzlica 31 no.6:735-738 Je\*63

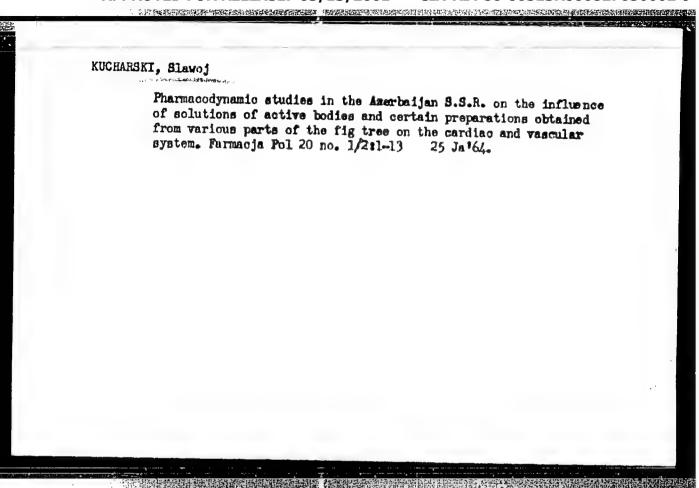
1. Wojewodzka Poradnia Przeciwgruzlica, Lublin.

#### KUCHARSKI, Ryssard

Generalised melanoma in a patient with pulmonary tuberculosis. Gruslica 28 no.12:1011-1017 D '60.

1. Z Kliniki Ptysjatrycznej A.M. w Lublinie, Kierownik: doc. dr H. Mysakowska.

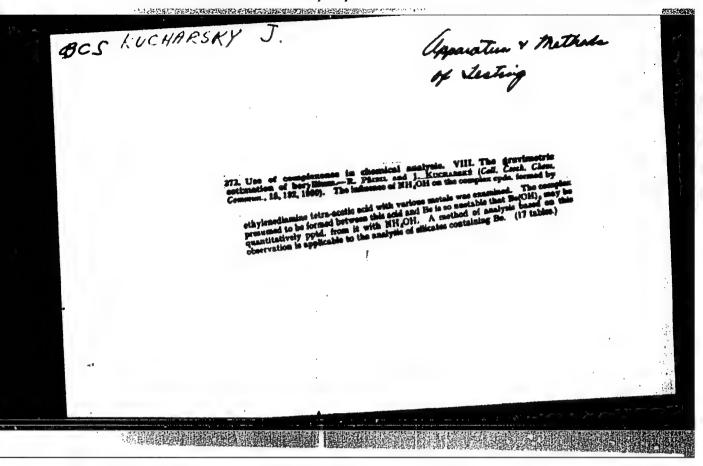
(TUBERCULOSIS PULMONARY compl) (MELANOMA compl)

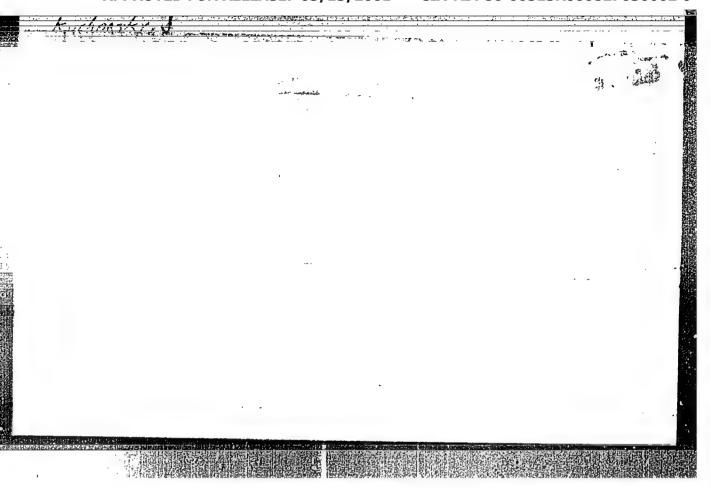


KUCHARSKI, Slavoj, st. asystent

In the thousand-tower city; rememirances of a fellowship holder of the Ministry of Health from his stay in Azerbaijan. Farmacja Pol 20 no. 3/4:119-124 25 F 164.

1. Akademia Medyczna, Poznan.





KUCHARSKY, Z.

TECHNOLOGY

Periodical: POZETNI STAVBY. Vol. 6, no. 10, Oct. 1958.

KUCHARSKY, Z.; SKUTA, R. Prefabrication of sanitary installations of the T 16 type in plants assembling structural elements. p. 601.

Monthly List of East European Accession (EEAI) LC, Vol. 8, no. 3
March 1959 Unclass.

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827030002-9"

# KUCHARZ, J.

The influence of state farms on individual farms, p. 856.

NCWE ROLNICTWO. (Panstwowe Wydawnictwo Rolnicze i Lesne) Warszawa, Poland. Vol. 8, no. 23, Dec. 1959.

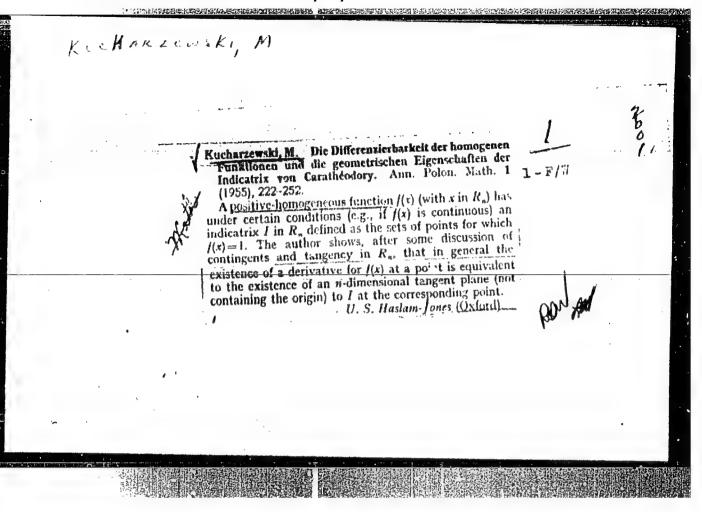
Monthly list of East European Accessions (EFAI) LC, Vol. 9, no. 2, Feb. 1960

Uncl.

EUCHARZEWSKI, Bohdan

A spectrographic method of analyzing metallic tungsten. Chem anal 7 no.2: 349-354 62

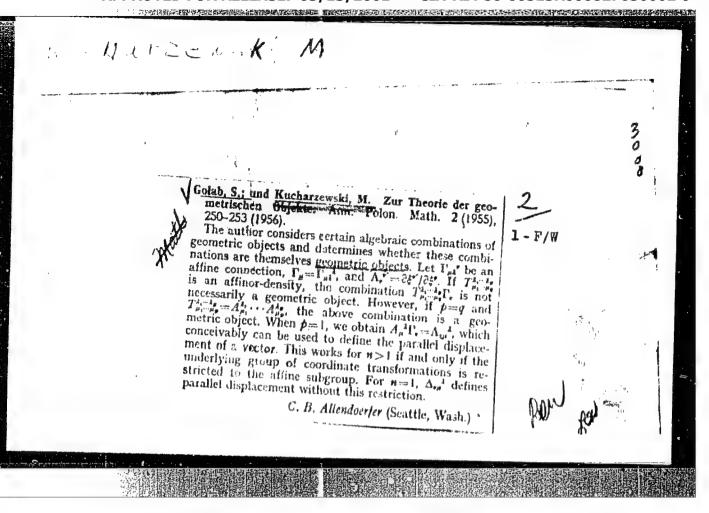
 Zaklad Chemii Analytycznej, Instytut Badan Jadrowych, Polska akademia Nauk, Warszawa. Kierownik Zakladu: prof. dr. J. Kinczewski.

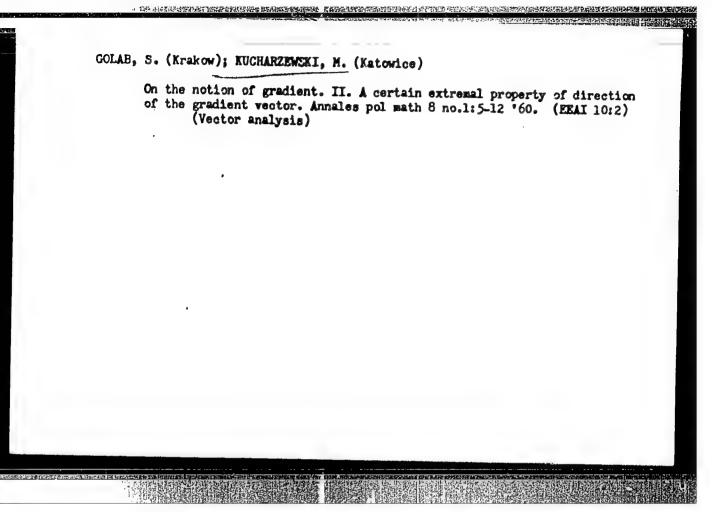


Generalization of the Euler equation for homogeneous functions. In Cerman. p.326
Abhabits and MCI (No. 2, 1955

Vol. 1, no. 2, 1955

So. East European Accessions Vol. 5, No. 9 September 1956

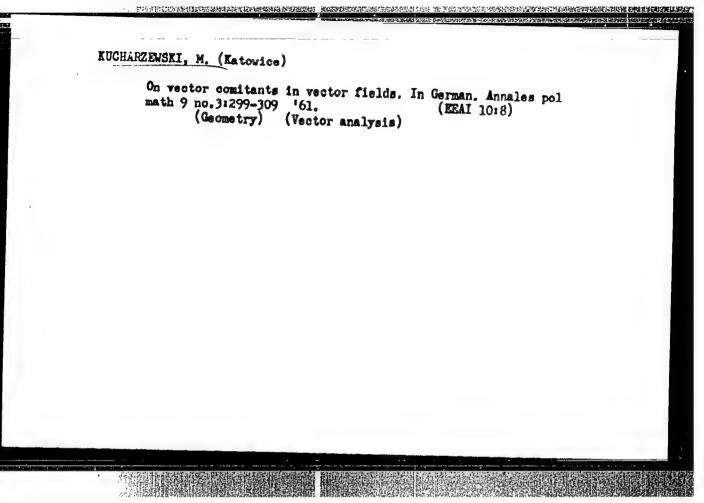


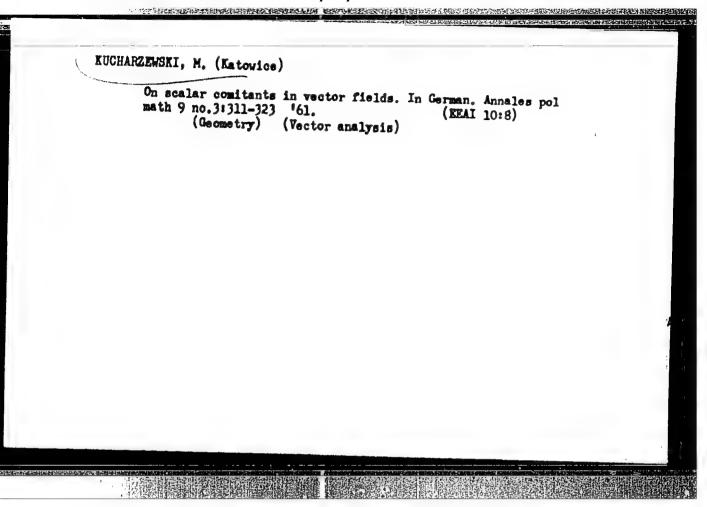


GOLAB, S. (Krakow); KUCHARZEWSKI, M. (Katowice)

Remark about the theory of comitants. Acta mat Hung 11 no.1/2:
173-174 '60. (EEAI 9:12)

1. Vorgelegt von G.Hajos.
(Numbers, Theory of)
(Calculus of tensors)

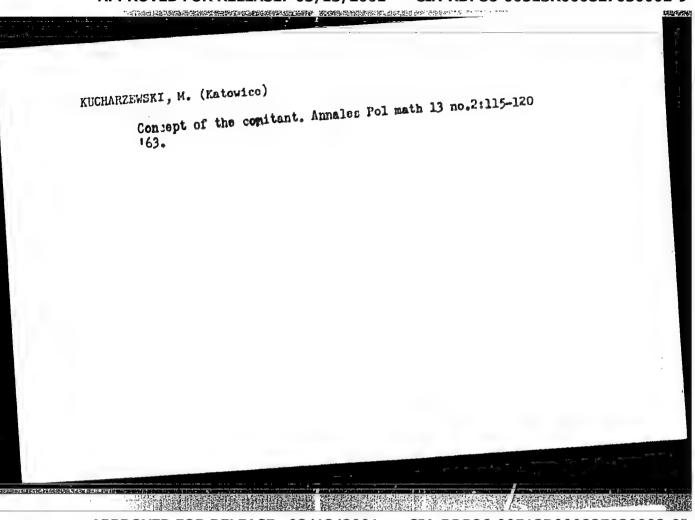




KUCHARZEWSKI, M. (Katowice); KUCZMA, M. (Krakow)

On the functional equation F(A B) F(A) F(B)

Annales Pol math 13 no.1:1-17 '63.



APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827030002-9"

KUCHARZEWSKI, M. (Katowice); KUCZMA, M. (Krakow)

Determination of geometric objects of the type [2,2,1] with a linear homogeneous transformation formula. Annales Pol math 14, no. 1:29-48 '63.

On a system of functional equations occurring in the theory of geometric objects. Ibid.:59-67.

GOLAB, S. (Krakow); JAKUBCWICZ, A. (Szczecin); KUCHARZEWSKI, M. (Katowice);
KUCZMA, M. (Katowice)

The geometric object representing a direction makes sense.
Annales Pol math 15 no.3:233-236 164.

KUCHARZEUSKi. M. (Katowice); KUCZMA, M. (Krakow.

Determination of linear differential geometric objects of the first class, with two components, in a two-dimensional space. Annales Pol math 40 no. 1:77-84.

KUCHARZEMSKI, M.; KUCZMA, M.

Basic concepts of the theory of geometric objects. Rosprany
matemat 43:1-72 '64.

# 26612

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P/032/60/007/003/002/002 D259/D301

Prosnak, WYodzimierz, J., and Kucharzyk, Piotr

(Warsaw)

AUTHORS:

TITLE:

On a profile with blowing aileron or flap

PERIODICAL: Archiwum budowy maszyn, v. 7, no. 3, 1960, 313 - 333

TEXT: This article was delivered by P. Kucharczyk at the 1959, September 7 - 13 conference on the Zaklaa Mechaniki Cieczy 1 September 7 - 13 conference on the Zaklad Mechaniki Cleczy 1
Gazow IPPT - PAN (PAS - IPPT Section of the Mechanics of Liquids and Gases), in Augustow. The purpose of the authors' work was and Gases), in Augustow. The purpose of profiles with blowing to examine the aerodynamic properties of profiles with blowing ailerons or flaps. Experiments carried out by the Katedra aerodynamiki political decomposite (Aerodynamics Department of dynamiki politechniki Warszawskiej (Aerodynamics Department of the Warsaw Polytechnic) on the flow around the airfoil with a jet flap only in the central section of the span, shows that an increase of the angle of attack appears also at the end section of the airfoil not directly affected by the jet. The effect of the jet preventing the separation of the boundary layers is ev-

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THE REPORT OF THE PROPERTY OF APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827030002-9" CALEGORIAN CONTRACTOR OF THE PROPERTY OF THE P

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On a profile with blowing...

ident only in the central section under the direct influence of the jet. It was further established that with a blowing jet, the separation of the boundary layers from the airfoil end sections occurs at smaller angles of attack than without the help of the occurs at smaller angles of attack than without the help of the jet. For their treatment of the problem, the authors assumed a jet. For their treatment of the problem, the authors assumed a steady flow of an ideal fluid. The airfoil profile and the aileron, as well as the mechanical flap, were represented by rectilinear sections and the jet sheet in the trailing edge of the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the profile proper. On the airfoil profile by a jet source on the graph and the airfoil profile and the airfoil

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 $c_y = f_1(\alpha, \beta, \delta_1, \delta_2, \delta_3,) + f_2(c_j, \alpha, \beta, \delta_1, \delta_2, \delta_3)$  (48) On a profile with blowing...

and the drag by formula

(44)

where d is the angle of attack;  $\beta$ , the flap angle;  $c_1$ , the flow coefficient,  $\delta_1$ ,  $\delta_2$ , and  $\delta_2$ , the geometrical parameters, the functions  $f_1$  and  $f_2$  are defined parametrically by mathematical calculation. There are  $\delta$  figures and  $\delta$  references: ematical calculation. There are  $\delta$  figures to the 3 Soviet-bloc and 2 non-Soviet-bloc. The references to the English-language publications read as follows: W. Prosnak and J Soviet-bloc and 2 non-soviet-bloc. The references to the English-language publications read as follows: W. Prosnak and Properties of the Influence of the Ground on the Aerodynamic P. Kucharczyk: The Influence of the Ground on the Aerodynamic Properties of an Airfoil with Jet Flap, "Archiwum Mechaniki Prosnak: Theory of Two-Dimensional Aerofoil with Jet Flap, W. Prosnak: Theory of Two-Dimensional Aerofoil with Jet Flap, "Archiwum Mechaniki Stosowanej", PWN, Warsaw 1958, vo. 10, no.

Card 3/4

P/032/60/007/003/002/002

D259/D301

On a profile with blowing...

1(3-24).

Katedra aerodynamiki Politechniki Warszawskiej (Aerodynamics Department of the Warsaw Polytech-ASSOCIATION:

nic Institute)

SUBMITTED:

February, 1960

Card 4/4

KUCHAVA, A.A.

Isolated pericardial wound. Khirurgiia no.10:80-81 0 '53. (MIRA 6:11)

1. Is fakul'tetskoy khirurgicheskoy kliniki (direktor - professor A.A.Kosyrev) Kubanskogo meditsinskogo instituta. (Pericardium--Wounds and injuries)

Paired and separate work of the cerebral cortex in dogs. Soob.All Gruz.SSR 28 no.1:89-96 Ja 162. (MJRA 15:4)

1. Akademiya nauk Gruzinskoy SSR, Institut fiziologii Tbilisi. Predstavleno akademikom I.S.Beritashvili. (CEREBRAL CORTEX)

BAGDAVADZE, N.V.; BARBAKADZE, L.V.; GINTURI, E.N.; KUCHAVA, N.Ye.;
MCGULISHVILI, L.M.; KHARABADZE, N.Ye.

Radioactivation method for determining gold in the blood. Soob.

AN Gruz. SSR 39 no.2:287-294 Ag '65.

(MIRA 18:9)

1. Institut fiziki AH GrumSSR. Submitted January 15, 1965.

BERDZENISHVILI, B.G.; VAYNSHIEYN, B.Z.; ZHITKOV, N.Ya.; KUCHAVA, V.A.;

Lightened pantograph for high-speed rolling stock. Elek. i
(MIRA 16:6)
tepl. tiaga 7 no.3:6-7 Mr 163.

1. Sotrudniki otdela elektricheskoy tyagi Nauchno-issledovatel skogo elektrotekhnicheskogo instituta Seveta narodnoge khezyaystva Gruzinskoy SSR. (Electric railroads---Wires and wiring)

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827030002-9"

KUCHAY, L.N.

THE PARTY

#### PHASE I BOOK EXPLOITATION

307/6030

Samsonov, G. V., Corresponding Member, Academy of Sciences UkrSSR;
A. T. Pilipenko, Doctor of Chemical Sciences, Professor; T. N.
Nazarchuk, Candidate of Chemical Sciences; O. I. Popova, Candidate of Chemical Sciences; and T. Ya. Kosolanova, V. A. Obolonchik, G. Kh. Kotlyar, L. H. Kuchay, V. P. Kopylova, G. T. Kabannik, A. Kh. Klibus, K. D. Hodylevskaya, and S. V. Radzikovskaya.

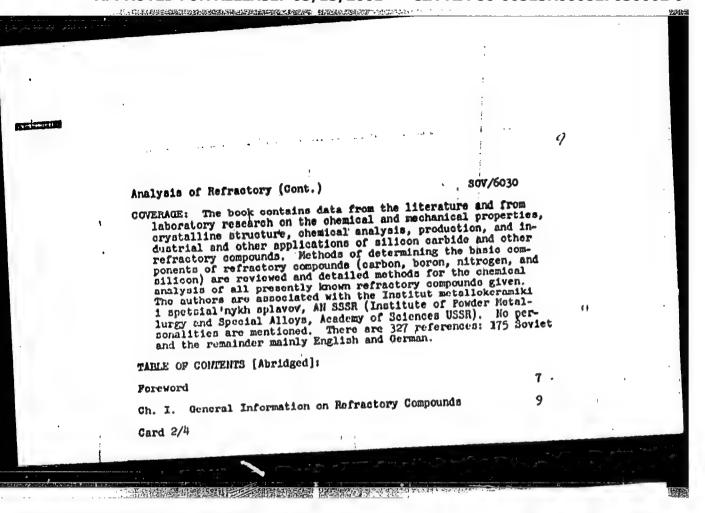
Analiz tugoplavkikh soyedineniy (Analysis of Refractory Compounds)
Moscow, Metallurgizdat, 1962. 256 p. 3250 copies printed.

Ed.: Yo. A. Hikitina; Ed. of Publishing House: O. M. Kamayeva; Tech. Ed.: A. I. Karasev.

PURPOSE: This book is intended as a laboratory manual for personnel in plant laboratories of the machinery, chemical, and aircraft industries and scientific research institutes. It can also be used by chemistry students at universities and schools of higher education.

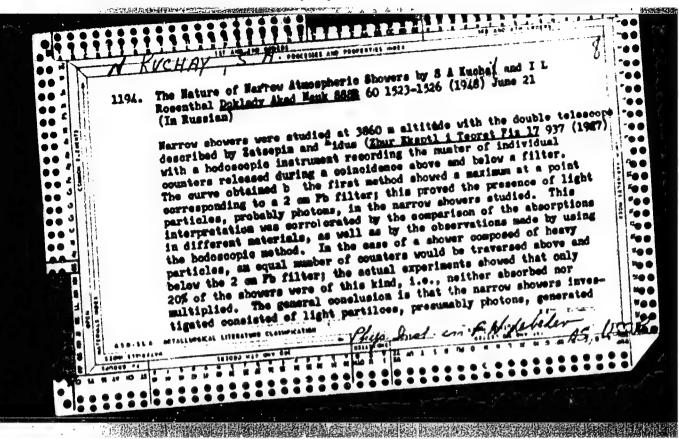
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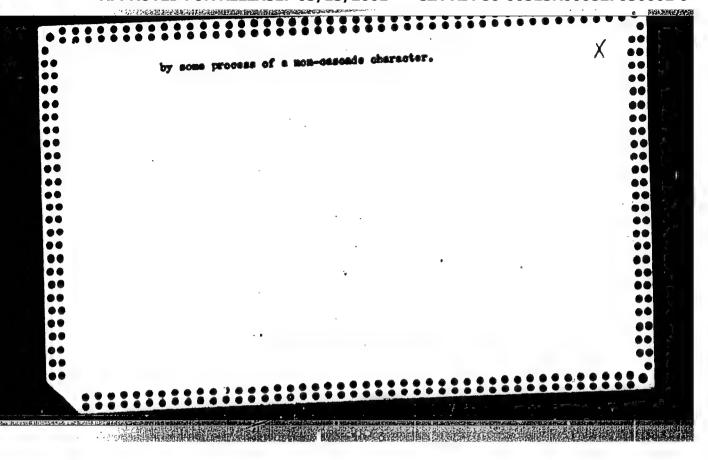
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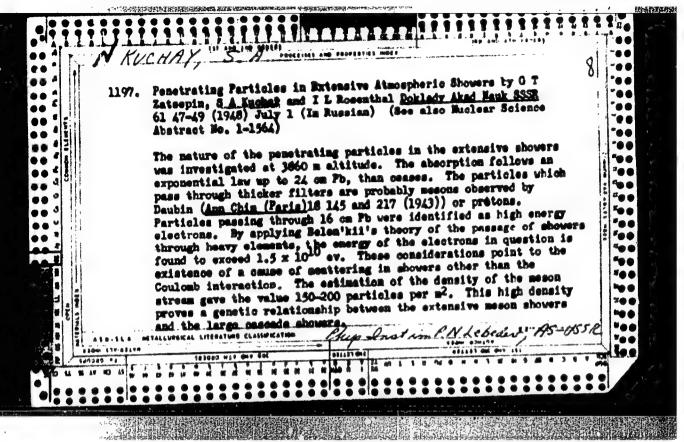


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Karbay, & D.

AUTHORS: Rodin, A.M. and Kuchay, S.A.

120-4-19/35

Measurements of the Depth of Penetration and the Coeffici-TITLE:

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ent of Diffusion of a Gas in a Metal (Izmereniye glubiny proniknoveniya i koeffitsiyenta diffuzii gaza v metalle)

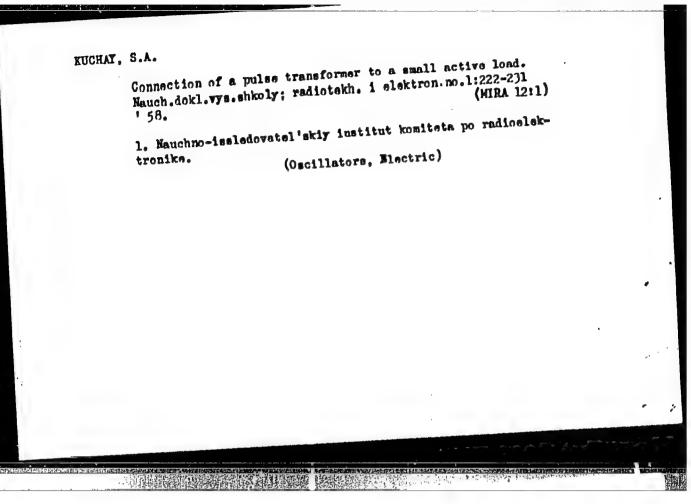
PERIODICAL: Pribory i Tekhnika Eksperimenta, 1957, No.4, pp. 68 - 69 (USSR).

ABSTRACT: The surface layer of a metal disc can be saturated with any gas by positive ion bombardment in gas discharges, or devices similar to a mass-spectrometer (Ref. 1-5), It is of interest to determine the depth of penetration L of the gas into the body of the metal and also the coefficient of diffusion D of the given gas in the metal. The one-dimensional diffusion equation is solved assuming that the coefficient of diffusion does not depend on the co-ordinates and the gas concentration, and subject to simple boundary conditions. The above assumption applies in a number of cases (Le Claire and Rowe, Ref. 6). There are 1 figure and 7 references, 2 of which are Slavic.

SUBMITTED: March 1, 1957.

Library of Congress AVAILABLE:

card 1/1



A TOPPOTENTIAL PROPERTY OF THE KUCHAY, S.A. 89-2-24/35 Ruchay, S. A., Rodin, A. M. The electric Absorption of a Gas by a metal with a Diffusing AUTHORS & Surface (blektricheskoye pogloshcheniye gaza metallom s ra-TITLE: apylyayushcheyaya poverkhnout'yu). Nr 2, pp. 202-205 (USSH) Atolanaya Energiya, 1958, When certain metals are bombarded with gas-ions whose energies PLAIGLIGAL: are in the order of magnitude of some to eV, an accumulation of gas-molecules is noticed in the metal. This is e.g. used in ABANRACT: isotopic separation of inert gases. When no chemical interaction between the gas and the metal starts, it is justified to assume that the gas-concentration is determined by its isotropic diffusion from the source which lies somewhat deeper than the range of the ions. From this follows that after a long irradiation the absorbed gas would have to be observed in all metallayers in comparable concentrations and that the entire quantity of gas would only be determined by the thickness of plate. It was experimentally determined, however, that this is not the case, but that the entire was accumulated in a layer, the thickness of which approximately corresponds to the range of ions. This contradiction can be removed by assuming that the dird 1/2

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39-2-2 /33 The electric Assorption of a Gas by a Metal with a Diffusing Bar.ace.

"electric absorption" is connected with a cathodic evaporation from the metal surface. On this assumption the electric absurption is theoretically calculated, where for the sake of simplicity the following assumptions are made: a) The ionic range in the metals is the same for all ions.
b) The thickness of the metal plate is great.

c) The diffusion coefficient is not dependent on the mordinates and the concentrations.

at first the solution of the diffusion equation is given and

then the following cases are treated in particular:

a) Steady distribution

b) Transition process in the source plane.

c) Gas concentration

There are 3 figures, 8 references, 2 of which are Slavic.

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3. Lactopes-2. Metals-Bombardment 1. Gases-Absorption Cord 2/2 Separation

	57 28-3-10/53
AUTHOR:	Kuchay, S. A. Stage of the Current
TITLE:	Concerning the Problem of the Initial Stage of the Current Passage Through a Diode (K voprosu o nachal'noy stadii pro- khozhdeniya toka cherez diod)
PERIODICAL:	Zhurnal Tekhnicheskoy Fiziki, 1958, Vol. 28. Kl. 3, Fr. (USSR)
ABSTRACT:	The transit-time of the front electrons under the action of a constant voltage, applied pulse-like to cylindrical and a constant voltage, applied pulse-like to cylindrical and a pherical diodes is evaluated here. The ratio of the transit-time of the -time of the front-electrons t to the transit-time of the electrons in a vacuum in the absence of a space charge - ty:
	electrons in a vacced unity . At first the lower limit of this
	ty cannot exceed and type considerations which ratio is evaluated starting from simple considerations which also apply to cylindrical and spherical diodes. For flat also apply to cylindrical diodes with infinite emission cylindrical and spherical diodes with infinite
Card 1/2	

57 20 3 10/33

Concerning the Problem of the Initial Stage of the Current Passage Through a Diode

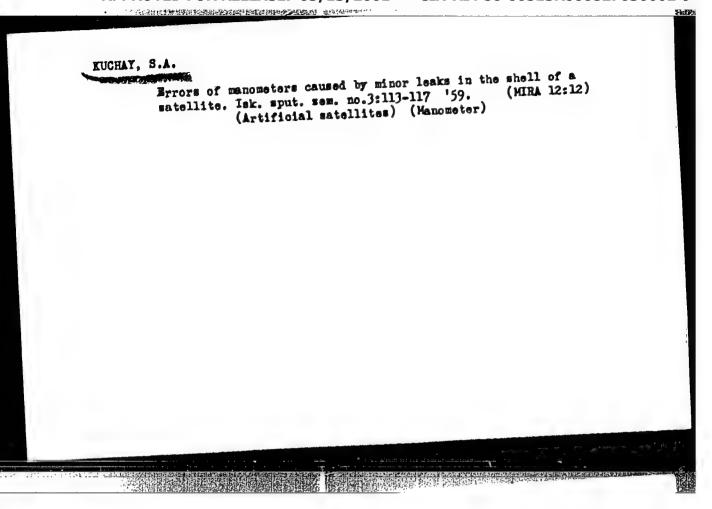
 $E_K \subset E_{\rm cath}$ , applies. I. e. in the case of the same position of the electron-front x the actual field  $E_K$  acting upon the front-electrons is always less than the field  $E_{\rm cath}$  at the cathode ( in an assumed diode whose cathode moves together with the electrons sitting on its surface). Therefore in a model with movable cathode the period t will be greater than the transit-time  $t_{\rm cath}$ . Equations (2) and (3) for the cylindrical diode with an anodic radius a and for a spherical diode with the same anodic radius, respectively are written down. There are 2 figures, and 2 Soviet references.

SUBMITTED: April 18, 1957

1. Diodes-Electrical properties 2. Mathematics

Card 2/2

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Double Focussing of a Beam of Charged Particles in a Large AUTHOR: Kuchay, S. A.

PERIODICAL: Pribory i tekhnika eksperimenta, 1959, Nr 4, pp 96-97

ABSTRACT: It is known (Refs 1-3) that the fringe field of a real magnet gives vertical focussing. However, when the magnet gap is small compared with the radius of curvature of the beam, Khurgin (Ref 1) showed that as far as the vertical effect on the beam is concerned, the boundary of a uniform field magnet is similar to a lens and T = ctg & (Fig 1). In a small gap the fringe field has very little effect on the horizontal focussing and in the first approximation one may assume that the uniform field abruptly stops on the boundary of the pole pieces. In this approximation the horizontal focussing condition is (Ref 4):

horizontal focussing condition is (Ref 4):

o sin  $\mathbb{D} + 1 \cos (\mathbb{D} - \varepsilon)/\cos \varepsilon = 0$ . The simultaneous solution of these two equations gives the following condition for double focussing:

 $1/\rho = (-\Phi \sin \Phi)/(\Phi \cos \Phi + \sin \Phi)$ .

Card 1/2 Thus, double focussing can be realized at least in a magnet

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Double Focussing of a Beam of Charged Particles in a Large Magnet

with a very small gap. However, in the case of a real magnet, the first approximation does not really hold and further calculations can only be carried out numerically. Fig 2 shows the vacuum chamber of a special model used in these numerical calculations (with the top lid off). In Fig 2, numerical calculations the magnet. Numerical calculations show that 3 represents the magnet. Numerical calculations show that double focusing is, in fact, possible even in a large gap. There are 3 figures and 5 references, of which 1 is Soviet, 1 German and 3 English.

SUBMITTED: July 16, 1958.

Card 2/2

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Koval'skiy, G.A. and Kuchay, S.A. AUTHORS:

Investigation of Small-Scale Ion Pumps

TITLE PERIODICAL: Pribory i tekhnika eksperimenta, 1960, Nr 2, pp 110-115 (USSR)

ABSTRACT:

An important problem in vacuum technology is the development of high-vacuum pumps, in which the extraction of gas from the working volume is not accompanied by the back migration into this volume of Only one type of pump the working fluid of the pump. is known at present which satisfies this requirement. The pump is based on the removal of gas which is first ionized, with the aid of electric and magnetic fields. However, the ion pump described in the literature (Ref 1) has a length of about 4 m and consumes 42 kW, the pumping speed being approximately 5000 litres/sec. therefore arises as to whether this particular design is the only possible one or whether other versions are

possible, in particular those in which the energy consumption is lower and the linear dimensions are

A series of experiments was carried out by the smaller.

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Investigation of Small-Scale Ion Pumps

authors in order to study the pumping action of a gasdischarge with a relatively short column in a magnetic field. These experiments did not lead to the development of the working model but nevertheless the results obtained are of definite interest and are now reported. shows the pumping system employed. The experiments were carried out on two installations in which the high-vacuum part of the discharge column was 5 cm long (apparatus A) and 12 cm long (apparatus B). Both A and B had identical vacuum systems and differed only in the design of the cathode. In A the cathode was directly heated and was in the form of a flat spiral, while in B the cathode was in the form of a rectangular plane surface, heated by electron bombardment. The cross-sections of the channels between the fore-vacuum and the high-vacuum regions were circular in A and rectangule in B. The discharge current was varied between 0 and 3A; the voltage between 100 and 500 V and the magnets produced a field of 2500 Oe in A and 5000 Oe in B. Pumping speeds between 14 and 46 litre/sec were obtained

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Investigation of Small-Scale Ion Pumps

for model B with fore-vacuum to high-vacuum pressure ratio between 14.5 and 5.5 respectively. The pumping speed achieved with model A was 1.7 litre/sec with the fore-vacuum to high-vacuum pressure ratio of 3.5. It was found that the pumping characteristics are improved when the magnetic field is increased. It was also found that there is an optimum discharge current at constant voltage (Fig 7). It was noted that an increase in the voltage across the discharge always improves the pumping characteristics. It is concluded that it is definitely possible to produce small-scale ion pumps working with a pressure drop of about 100. There are 7 figures and 1 English reference.

SUBMITTED: February 26, 1959

Card 3/3

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77300 50V/57-30-2-3/18

AUTHOR:

Kuchay, S. A.

TITLE:

Optical Properties of Axially Symmetrical Magnetic Fields With a Central Source of Charged Particles

PERIODICAL:

Zhurnal tekhnicheskoy fiziki, 1960, Vol 30, Nr 2,

pp 142-152 (USSR)

ABSTRACT:

Axially symmetrical magnetic fields are often used for magnetic focusing or for mass analysis of charged particles. The most used fields are those

with mirror symmetry:

 $H_r(r, z) = -H_r(r, -z),$  $H_s(r, z) = H_s(r, -z).$ 

where the field components are in the cylindrical coordinate system. The author denotes trajectories lying in the central plane z=0 as plane trajectories. In a given field H(r,0) particles with

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same value of the parameter  $[H \rho]$  = pc/e

Optical Properties of Axially Symmetrical Magnetic Fields With a Central Source of Charged Particles 77306 804/57-30-2-3/18

have a plane stationary orbit r = r, if the condition:

 $H(r_e)r_e = !H_V$ .

holds on the circle of  $r=r_c$ , z=0. The author limits his inquiry to fields containing only one stationary orbit and only to fields with a source central with respect to the stationary orbit. The latter are called circular fields. Shpinel pointed out (DAN SSSR, 53, 801, 1946) that there exist nonuniform fields in which a particle with a critical value of  $[H \ \rho]$  ac describes a spiral trajectory approaching the stationary orbit asymptotically from the inside, while the particles with  $[H \ \rho]$  ac cross the orbit under a

[H $\rho$ ] > [H $\rho$ ] ac cross the orbit under a finite angle and never return to it. Such fields are, therefore, called fields with a cutoff. In a

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Optical Properties of Axially Symmetrical Magnetic Fields With a Central Source of Charged Particles 773-5 507/57-30-2-3/18

later paper Shpinel! (ZhTF, XX, 834-846, 1950) computed the plane trajectories and described a

orbit. In 1953 Korobochkin noticed that the field with a cutoff can be represented as a limiting case of a more general relationship between the radial distribution of the circular field and its optical properties. In the present paper the author gives an interpretation of the equations of motions in axially symmetrical fields, describes their optical properties in the central plane, and the spacial motion in circular fields. He shows the way for using the circular fields as a mass-separating device with a source extended along the Z axis. The field can be represented by means of the vector potential A # Ap , given by:

 $A(r, z) = \frac{1}{r} \int_{0}^{r} H_{r}(\zeta, z) \zeta d\zeta. \tag{2}$ 

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Optical Properties of Axially Symmetrical Magnetic Fields With a Central Source of Charged Particles

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whenever the transition to the limit of  $\rightarrow$  0 is possible. The author writes the equations of motions as:

 $m! = \frac{\partial}{\partial \xi} \psi(r, z) |\xi - r, z,$ 

where

$$\psi(r, z) = \frac{p^3}{2m} \left\{ 1 - \left[ \frac{A}{|H_p|} - \frac{r_0}{r} \left( \frac{A_0}{|H_p|} - \cos \alpha_0 \right) \right]^2 \right\},$$

where  $\alpha_0$  = angle between the initial velocity and the  $r=r_0$  parallel;  $z=z_0$ , passing through the origin. The function  $\psi$  corresponds to the kinetic energy of the meridional motion of the particle, while the cosine of the angle between the velocity at a given point and the parallel through that same point is given by:

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77306 SOV/57-30-2-3/18

$$\mathfrak{t} = \frac{A}{|H_0|} - \frac{z_0}{r} \left[ \frac{A_0}{|H_0|} - \cos \alpha_0 \right] \tag{3}$$

At this point the author introduces the concept of the nonhomogeneity index k given by:

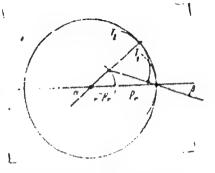
$$k = \lim_{\substack{\alpha,\beta \to 0 \\ \text{sin } \beta}} \frac{\sin \alpha}{\sin \beta} = \frac{\rho_r}{r_r - \rho_r}. \quad (6)$$

The quantities used are shown in Fig. 2. The index depends not only on the field distribution but also on the value of the [H  $\rho$ ] parameter, since the field has, in general, different optical properties for beams of various impulses. For a uniform field k has the limiting value of 1. The author develops equations for the field at the point of return  $r_r$ , the value of  $(d\xi/dr)_{r=r_r}$ , and the azimuth of the point of return  $r_r$ . Using the

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Fig. 2. Determination of nonhomogeneity index k.  $T_1$ ,  $T_2$  are arcs of trajectories with same H near point of turn.



parameter  $\mu=r/r_p$  he discusses the function  $\xi$  (  $\mu$  ) and plots some examples. He shows that the case  $\rho_r=r_p$  corresponds to the case when the trajectory approaches  $r_p$  asymptotically, and one is dealing with a field with a cutoff. In this case

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 $1/k \rightarrow 0$ , and this is then the other limit of the k values. Next, the author derives a formula for the aberration:

$$\Delta_{\mathbf{y}} = \frac{r_{\mathbf{y}} \, \mathbf{y}^3}{2k}$$
 .

and for the linear magnification gives an approximate expression:

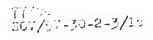
$$\Delta_{r} = k r_{er} \tag{17}$$

All symbols used are shown geometrically on Fig. 6. Finally, the mass dispersion is given by:

$$\Delta_m = \frac{kr_T}{2} \frac{km}{m} +$$

Card 7/12

It follows that for a given k all focusing properties of the central plane can be expressed directly using



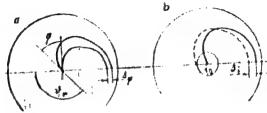


Fig. 6. Symbols used in equations of optical properties in the central plane.

the nonhomogeneity index. The collectors of isotopes located at  $\mathcal{A} = \mathcal{A}_r$  must have a clearance given by:

 $s = \Delta_m - \Delta_j - \Delta_r = kr_r \left( \frac{1}{2} \frac{km}{m} - \frac{r_0}{r_r} - \frac{1}{2k^2} \varphi^2 \right).$ 

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77306 SOV/57-30-2-3/18

One sees that separation capabilities of circular waves increase with increasing k. The author nevertheless points out that this simple picture is true only for the central plane. Other noncentral trajectories with impulses close to the critical trajectories with impulses close to the critical oscillate around the central plane but cross it comparatively far from r = r, This makes difficult

the use of circular field for separation purposes. To investigate the matter further, the author computed numerically some trajectories for the case of finite-sized sources. He shows the field due to pole on Fig. 5(only right side of fig. shown).

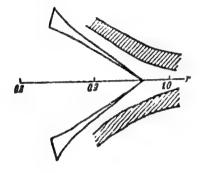
Fig. 5. Pole shape of the model.



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He shows on Fig. 9 the form of the trajectory projection "lines" for an infinitely thin source and a collector at  $\theta$  = 2, the angular dispersion of the beam being  $\varphi = \pm 10^{\circ}$ .



Card 10/12

Caption to Fig. 9 on Card 11/12

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Fig. 9. Shape of lines of a cutoff field. Infinitely thin source is  $0.2r_p$  high. The collector is a meritional plane with an azimuth  $\sqrt{r} = 2$ . Cross-hatched region is part of the heavy isotope "line" with  $\sqrt{r} = 10\%$ . On the r-axis are divisions in  $r_p$  units.

Poor separation properties of the fields with a cutoff are due to the sharp rise of vertical forces near the point of turning back. This may be avoided by computing the field and the operating values of

THO I in such a way that the separation takes place for smaller values of k. Although this reduces the clearance in the central plane, one gets a general improvement in the picture. The author believes that circular fields can be used for mass separation in the case of divergent beams. In

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Optical Properties of Axially Symmetrical Magnetic Fields With a Central Source of Charged Particles 77306 \$07/57-30-2-3/18

view of their considerable dispersive power, one does not have to care about the removal of aberration, about the reduction of the unavoidable divergence of the beam, or the ultimate dimension of the source. There are 11 figures; and 8 references, 2 Soviet, 2 Japanese, 2 German, 1 U.K., 1 U.S. The U.K. and U.S. references are: E. Rae, Phil. Mag., 41, 525-533, 1950; N. Coggeshall, M. Muskat, Phys. Rev., 66, 187, 1944.

SUBMITTED:

Card 12/12

#### CIA-RDP86-00513R000827030002-9 "APPROVED FOR RELEASE: 03/13/2001

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KUCHAYEY, V.L.

USSR/Electronics - Electrical effects

Card 1/1

Fub. 86 - 12/36

Authors

1 Kowrishnykh, O. H., and Kuchayev, V. L.

Title

Radioactive sources of high voltage

Periodical : Priroda 44/6, 86 - 89, Jun 1955

Abstract

An apparatus is described which consists basically of two plates, one grounded and the other ideally insulated and coated with a radicactive substance, the whole being placed in a vacuum. The principle on which a potential difference is developed is explained, such difference amounting in a given instance to 6,600 volts. Figures of quantities involved are stated. Pive English-lanuuage references (1913-1953). Drawings; graphs.

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#### "APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827030002-9

20 119 2-31/60 Boreskov; G. K., Kuchaye .. V. L. AUTHORS: The Catalytic Activity of Germanium in Relation to the 'TITLE: Isotopic Exchange Reaction Between Hydrogen and Deuterium (Kataliticheskaya aktivnost: germaniya v otnoshenii reaktsii izotopnogo obmena rodoroda s deyteriyem) Doklady Akademii Nauk SSSR, 1958, Vol. 119, Nr. 2, PERIODICAL: pp/ 302-304 (USSR) The present paper compares the specific catalytic activity ABSTRACT: of a germanium semiconductor element with the activity of the transition metals having incompletely filled d-zones. This comparison is here made with respect to the reaction of isotopic exchange of hydrogen with deuterium. The authors investigated the catalytic activity of germanium by means of the static method with circulation. The content of HD in the hydrogen-deuterium mixture was determined by means of the method of thermal conductivity. The reaction took place in a reaction vessel of quartz within the temperature interval 350° - 550°C, the equimolar Card 1/4

The Catalytic Activity of Germanium in Relation 20-119.2-31/60 to the Isotopic Exchange Reaction Between Hydrogen and Deuterium

mixture of hydrogen with deuterium having had a pressure of 90 - 190 mm torr. Monocrystalline germanium with an electronic line of the resistance 6 Ohm.cm served as catalyst. The gases hydrogen and deuterium used for reaction were produced electrolytically. The formula used for the calculation of the specific catalytic activity of germanium is given; it is valid for random mechanisms of the exchange of hydrogen and deuterium. A diagram shows the dependence of the catalytic activity of the two investigated germanium samples on the inverse temperature. The activation energy of the reaction amounted to 17 kgal/g-mol. The specific catalytic activity of the two germanium samples amounted to 3300~3.10-10 g-mol/cm2.sec. The catalytic activity of the samples determined at 650°C was a little greater. The reaction order was investigated with one of the two germanium samples at 480°C. The same degree of transformation at various pressures shows that the reaction takes place

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 The Catalytic Activity of Germanium in Relation 20-119-2-31/60 to the Isotopic Exchange Reaction Between Hydrogen and Deuterium

according to the first order. A table contains the specific catalytic activities of germanium and of some metals in relation to the reaction of the oxygen-deuterium exchange at 300°C and at a pressure of the mixture of 40 torr. In the experiments discussed here the exchange takes place according to the absorption-desorption mechanism when the surface of germanium is only little filled and when the absorption is the limiting stage of reaction. Im the transition to a stronger filling of the germanium surface the activation energy of the reaction must obviously increase and approach the desorption energy of hydrogen (about 41 kcal/g-mol). The catalytic activity of the metals of period IV increases with growing atomic number and reaches a maximum with nickel. The catalytic activity decreases strongly in the transition from nickel to copper. There are 1 figure, 2 tables and 2 references, 1 of which is Soviet.

Card 3/4

#### "APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827030002-9

. The Catalytic Activity of Germanium in Relation 20-119-2-31/60 to the Isotopic Exchange Reaction Between Hydrogen and Deuterium

ASSOCIATION: Nauchno-issledovatel'skiy fiziko-khimicheskiy institut

im. L. Ya. Karpova (Physico-Chemical Scientific Research Institute imeni L. Ya. Karpov)

PRESENTED: October 9, 1957, by A. A. Balandin, Member, Academy of

Sciences USSR

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SUBMITTED: Ootober,1, 1957

Card 4/4

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S/195/60/001/003/004/013 B013/B058

AUTHORS:

Kuchayev, V. L., Boreskov, G K

TITLE:

Isotopic Exchange of Hydrogen on Germanium Samples of

the n- and p-Type

PERIODICAL: Kinetika i kataliz, 1960, Vol. 1, No. 3, pp. 356 - 364

TEXT: In this paper the authors studied the effect of semiconductor properties of crystalline germanium on its catalytic activity with respect to the isotopic exchange of hydrogen with deuterium and the chemosorption of hydrogen, on germanium samples. Hydrogen and deuterium were produced by electrolysis. Germanium monocrystals crushed in vacuum, from which 5 samples with various resistivity and various type of conductivity were prepared, served as catalysts. The catalyst surfaces were measured directly in the reaction vessel after the adsorption of spectrally pure krypton at the temperature of liquid nitrogen. The calculation was made according to the BET method. The relative measuring accuracy amounted to about 5%.

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Isotopic Exchange of Hydrogen on Germanium Samples of the n- and p-Type

**3/**195/60/001/003/004/013 B013/B058

Table 2 shows the change of the germanium-sample surfaces under the influence of various temperatures. The catalytic activity of germanium with respect to isotopic exchange in homomolecular hydrogen was studied by the static method at pressures of the equimolecular hydrogen-deuterium mixture of 0.7 and 0.1 mm Hg at from 180 to 280°C. The calculation method was described in the paper by M. A. Avdeyenko, G. K. Boreskov, and M. G. Slin'ko (Ref. 9). No noticeable difference in the catalytic activity was ascertained between samples of different type of conductivity. The rate of adsorption of hydrogen was studied at room temperature (Table 3) and at 100°C before testing the catalytic activity. It was determined that rates of adsorption, energy of activation, and the adsorption as a function of the surface occupation are almost similar for all 4 samples. The adsorption isotherms of hydrogen were recorded at 210°, 244°, and 274°C and don't show any noticeable differences. With an occupation of up to 0 = 0.15, they correspond to Langmuir's equation for adsorption with dissociation. The heat of adsorption is 25 kcal/mole A deviation from Langmuir's equation and a lower heat of adsorption are to be observed at

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Isotopic Exchange of Hydrogen on Germanium Samples of the n- and p-Type

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a higher degree of occupation. The solubility of hydrogen in germanium is small and of no effect on the measuring results. As can be seen from the dependence of the rate of desorption on the surface occupation at 180°C (Fig. 10), it drops to one third after removal of about 20% of the adsorbed hydrogen from the germanium surface. It was established that the rate of desorption of the hydrogen-deuterium mixture is about 1.5 times greater than the rate of isotopic exchange, and about 1.5 times smaller than the rate of desorption of hydrogen, under equal conditions. The studies gave the following conclusive results: The rates of hydrogen adsorption and isotopic exchange were almost equal for all samples studied inspite of a change of the concentration of free electrons and holes by 7 to 9 orders of magnitude. This permits the conclusion that the adsorption of hydrogen on germanium proceeds without participation of free electrons or holes, i. e. without surface charge. Similar rates of desorption and isotopic exchange point towards an adsorption-desorption mechanism of the reaction. V. M. Frolov, O. V. Krylov, and S. Z. Roginskiy are mentioned. There are 10 figures, 3 tables, and 18 references: 5 Soviet, 10 US, 1 Dutch, and 3 German.

Card 3/6 .....

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Isotopic Exchange of Hydrogen on Germanium Samples of the n- and p-Type

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ASSOCIATION:

Fiziko-khimicheskiy institut im. L. Ya. Karpova (Physicochemical Institute imeni L. Ya. Karpov)

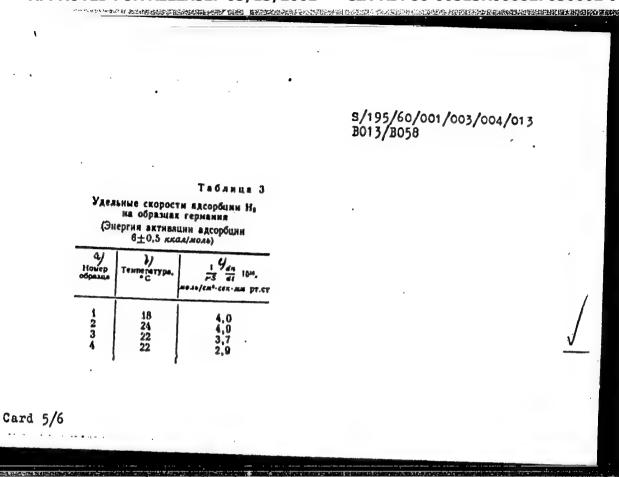
SUBMITTED:

June 27, 1960

il Howep obpasua	р Предварии жиля темперитурная обработка при			
	100*	3000	500*	630=
1 2 3 4 5	2,3 2,0 1,8 2,1*	1,9 1,6° 1,5° 1,8°	1,4 1,1° 0,05	1,3*

Card 4/6

## "APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827030002-9



\$/195/60/001/003/004/013 B013/B058

Logend to Table 2: The surface of the Ge samples, cm<sup>2</sup>. a) No. of the sample; b) heat treatment at:
Legend to Table 3: The specific rate of adsorption of  $\rm H_2$  on Ge samples (energy of activation  $\rm 6\pm0.5~kcal/mole)$  a) No. of the sample; b) temperature,  $\rm ^{O}C$ ; c) (1/PS) (dn/dt)·10<sup>14</sup>, mole/cm<sup>2</sup>·sec. mm Hg

Card 6/6

KUCHAYEV, V.L.; BORESKOV, G.K.

Relationship between the catalytic activity and semiconductor properties of germanium. Probl. kin. i kat. 10:108-110 '60.

(MIRA 14:5)

1. Fiziko-khimicheskiy institut imeni L.Ya. Karpova. (Germanium)

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"APPROVED FOR RELEASE: 03/13/2001
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                                                                                                                           5/020/62/145/004/022/024
                                           Kuchayev, V. L., and Boreskov, G. K., Corresponding Member
                                                                                                                            B101/B138
                                               Catalysis and adsorption of hydrogen on germanium films
                                                 Akademiya nauk SSSR. Doklady, v. 145, no. 4, 1962, 853-856
                 TEXT: The authors tested the data of Y. L. Sandler, M. Gazith (J. Phys. gauttered Ge films. Chem. 63. 1095 (1959)) on the datalytic activity of sputtered Ge.
                  TEXT: The authors tested the data of Y. L. Sandler, M. Gazith (J. Phys. films. Chem., 63, 1095 (1959)) on the catalytic activity of sputtered the in Special devices (Fig. 1) prevented contact between the Ge and the Special devices (Fig. 1) prevented its catalytic effect by immersion in tunketen wire coil and eliminated its catalytic effect.
           AUTHORS:
                    Special devices (Fig. 1) prevented contact between the Ge and the inmersion in tungaten wire coil and eliminated its catalytic effect by immersion mm tungaten wire coil and eliminated to 450°C and evacuated to 3.10-7 mm liquid Ga. The amoule was heated to
                     tungaten wire coil and eliminated its catalytic effect by immersion in He, and Ga. The ampoule was heated to 450°C and evacuated courrent (13.5 a) before sputtering on the Ga.
              TITLE:
                      liquid Ga. The ampoule was heated to 450°C and evacuated to 3.10-7 mm HE, (13.5 a)

liquid Ga. The ampoule was heated to 450°C and evacuated to 3.10-7 mm HE, (13.5 a)

This was done by passing a current (13.5 a)

This was done by passing a current (13.5 a)

Indicate on the Ge. ampoule immersed in liquid nitrogen.

The ampoule was heated to 450°C and evacuated to 3.10-7 mm HE, (13.5 a)

The ampoule was heated to 450°C and evacuated to 3.10-7 mm HE, (13.5 a)

The ampoule was heated to 450°C and evacuated to 3.10-7 mm HE, (13.5 a)

The ampoule was heated to 450°C and evacuated to 3.10-7 mm HE, (13.5 a)
                PERIODICAL:
                        through the W coil, with the ampoule immersed in liquid nitrogen. Tate according to the catalytic activity and rate of H2 adsorption were measured according to the catalytic catalytic activity and rate of H2 (1960). Results: (1) The catalytic catalytic is kataliz, 1, no. 3, 356 (1960). Results: (2) The catalytic catalytic activity and rate of H2 (29 - 56). 10-14 mole/cm<sup>2</sup>. sec. mm Hg. 15
   S
                            activity of H = D exchange at 100°C was less than 1.10 15 mole/on
  ři
  fil
  WW
 Wire
 in_1
 (3) (
for f
                                                                                                         the W coil; (5) copper wire
                                                                                      (6) glass rod operated by iron
bar (;
                              card 1/8
                                                                       -moving air bubbles from (3).
Card 2/
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                                                                                                                       CIA-RDP86-00513R0008270300
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Catalysis and adsorption of ...

S/020/62/145/004/022/024 B101/B138

i.e., in agreement with K. Tamaru, M. Boudart (Adv. in Cat., 9, 699 (1957)) and five times less than the values of Sandler and Gazith. This is attributed to the fact that these scientists had not eliminated the catalytic activity of the W coil. There are 3 figures and 1 table.

ASSOCIATION: Fiziko-khimicheskiy institut im. L. Ya. Karpova (Physicochemical Institute imeni L. Ya. Karpov)

SUBMITTED: April 29, 1962

Fig. 1. (A) Reaction vessel for studying the rate of H<sub>2</sub> adsorption on Ge films. (1) ampoule; (2) Mo lead-in, 3.5 mm diam; (3) coil of 0.5 mm wire; (4) graphite orucible; (5) glass tube; (6)~0.1 g Ge; (7) iron wire which is removed with a magnet. (3) Device for immersing the W coil in liquid Ga. (1) ampoule; (2) iron bar with which the test tube (3) containing liquid Ga (4) is raised toward the W coil; (5) copper wire for fixing the device in raised position; (6) glass rod operated by iron bar (7) sealed in glass, for removing air bubbles from (3).

Card 2/8 >

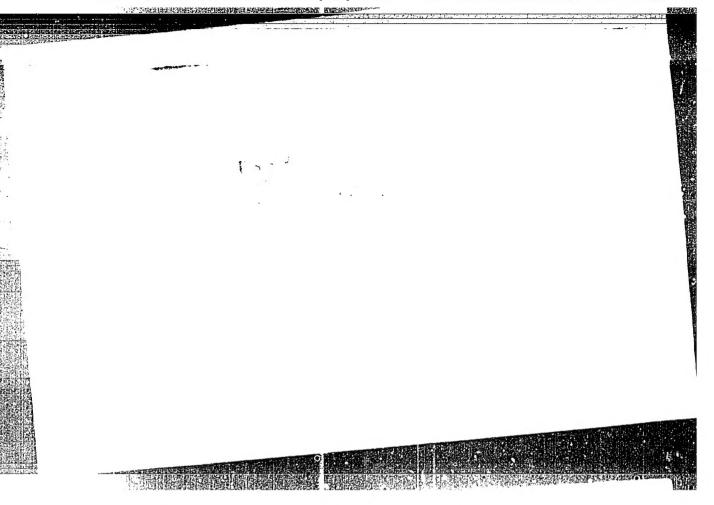
ACHOLOVA, G., KUCHAYEVA, A.

Distribution of actinoxycetes antagonists to the causative agents of outton diseases in Storczem acils of Uzbekistan.
Uch. biol. zhur. 9 no.1:19.22 165. (MIRA 18:6)

1. Institut botaniki AN U.SSR.

# "APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827030002-9

\*\*Microbes - Antagoniats in Plant Diseases," a paper presented at the Antibiotics Research Conf., Peiping, 1-6 December 1955
In library DB-38431



APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827030002-9"

KUCHAYEVA, A.G.; YEGGROVA, S.A.

Penetration of antibiotics into perennial plants. Mikrobiologica (MIRA 8:7)
24 no.3:315-320 My-3e '55.

1. Institut mikrobiologica Akademic nauk SSSR, Moskva.

(AITIBIOTICS, effects, on plants, penetration in perennial plants)

(PLANTS, effect of drugs on, antibiotics, penetration in perennial plants)

antibiotics, penetration in perennial plants)

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